## ANION-CONTROLLED REDISTRIBUTION REACTIONS IN THE SYSTEM $C_3C_1_3^+A^-/(Ch_3)_3Sin(Ch_3)_2^{-1}$

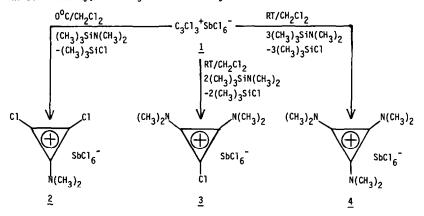
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<u>Summary:</u> Transamination reactions of the 4z-exchange type between  $C_3Cl_3^+A^-$  and  $(CH_3)_3Sin(CH_3)_2$  are introduced as a novel mode of aromatic substitution at the  $C_3^+$ -core. The extent of transamination can be controlled by appropriate choice of counterion  $A^-$ .

While many ligand redistribution reactions of the 4z-exchange type are known for trigonal boron systems  $^{2}$ ), virtually nothing is known about comparable reactions of isoelectronic carbeniumions. As a model case we have examined the reaction of silylamines with chlorocyclopropenium systems as their hexachloroantimonate-V salts and we now wish to report on some remarkable observations which we have made in the course of this study.

Thus we have found that  $C_3Cl_3^+SbCl_6^{-3}$  undergoes very clean transamination reactions with  $(CH_3)_3Sin(CH_3)_2$ , the extent of which can be controlled by simple variation of temperature and stoichiometry, according to the following scheme:



Under the reaction conditions given in the scheme partially aminated cyclopropenium systems  $\underline{2}$  and  $\underline{3}$  can be obtained in reasonable to good yield. It is noteworthy, that  $\underline{2}$  and  $\underline{3}$  are both inaccessible by the more conventional transamination technique of reacting  $C_3Cl_4$  with  $(CH_3)_2NH$ <sup>4</sup>.

2 represents the first monodialkylamino-dihalogeno- $C_3^+$ -system. In view of its two exchangeable Cl-substituents it is of potential value for generating new substitution patterns of the  $C_3^+$ -core. 2 is colorless and readily soluble in  $CH_2Cl_2$ , from which it must be precipitated immediately after formation with n-hexane (yield: 50%; mp.:  $180^{\circ}C$  (dec.);  $^1H$ -NMR (CDCl $_3$ ):  $\mathcal{T}=6.67$  (s); IR (KBr): 1930 (m), 1613 (s), 1428 (s), 1354 (s) cm $^{-1}$ ). Gradually in  $CH_2Cl_2$ -solution and almost instantanously in  $CH_3CN$  2 reacts with itself in what seems to be a disproportionation reaction (1) which produces bright yellow insoluble 3 in 80% yield

(mp.:  $252^{\circ}$ C (dec.); <sup>1</sup>H-NMR (CD<sub>3</sub>CN):  $\mathcal{T} = 6.78$  (s, 6H), 6.81 (s, 6H); IR (KBr): 1955 (m), 1635 (s), 1409 (s), 1390 (s) cm<sup>-1</sup>):

$$\frac{2}{\text{solution}} \rightarrow 0.5 \ \underline{3} \downarrow + 0.5 \ \underline{1}$$

The mechanism of this unprecedented ligandredistribution which formally takes place between two positively charged species is unclear at present.  $\underline{3}$  is also abtained as bright yellow needles in 65% yield by reacting  $\underline{1}$  with 2 moles of silylamine (cf. scheme) at room temperature and subsequent recrystallization of the crude product from  $CH_2CN$ .

So far synthesis of bisdialkylamino-chloro- ${\rm C_3}^+$ -systems via  ${\rm C_3Cl_4/R_2NH}$  has been restricted to such cases, where bulky groups R prevented triple amination  $\frac{4}{1}$ .  $\frac{3}{2}$  contains the hitherto inaccessible parent of the series, with a reactivity pattern quite different from the sterically hindered homologues  $\frac{5}{1}$ .

Further reaction of  $\underline{3}$  as a suspension in CH<sub>2</sub>Cl<sub>2</sub> with excess (CH<sub>3</sub>)<sub>3</sub>SiN(CH<sub>3</sub>)<sub>2</sub> gave the peraminated salt  $\underline{4}$  as highly insoluble dark violet cristals in virtually quantitative yield (mp.:  $182^{\circ}$ C (dec.);  $^{1}$ H-NMR (CD<sub>3</sub>CN):  $\mathbf{T}$  = 6.94 (s); IR (KBr): 1560 (s), 1403 (s) cm<sup>-1</sup>). As colour and solubility of this material differed so dramatically from the corresponding known perchlorate<sup>4</sup>) (which is colourless and very well soluble in CH<sub>2</sub>Cl<sub>2</sub>) a single crystal X-ray determination of  $\underline{4}$  was undertaken<sup>6</sup>). The results of this study fully confirmed the assumed structure and allowed to attribute the abovementioned properties of  $\underline{4}$  to a new kind of CT-interaction between cation and anion of this salt. A detailed discussion will be presented in a forthcoming full paper<sup>7</sup>).

Comparison of the  $^1\text{H-NMR}$  data of salts  $\underline{2-4}$  reveals a progressive down-field shift of methyl resonances in the sequence  $\underline{4}$  -  $\underline{3}$  -  $\underline{2}$  as an indication of the increasing electron demand of the  $C_3^+$ -core along this series. In addition,  $\underline{3}$  exhibits rotational hindrance around the C-N bonds, as is expected of a bisdialkylamino- $C_3^+$ -system. The infrared data of  $\underline{2-4}$  are characteristic of cyclopropeniumions of  $C_{2v}$  and  $D_{3h}$  symmetry, respectively  $^4$ .

Remarkably, replacement of SbCl $_6^-$  in  $\underline{1}$  by Cl $_1^+$ , BF $_4^-$  or CF $_3$ SO $_3^-$  8,9) resulted in ex-

Remarkably, replacement of  $SbCl_6$  in 1 by  $Cl_7$ ,  $BF_4$  or  $CF_3SO_3$   $^{8,9}$  resulted in exclusive <u>triple</u> amination upon reaction with  $(CH_3)_3SiN(CH_3)_2$ . Thus, the selective mono- and disubstitution processes reported above are due to a novel type of counterion control  $^{10}$ , the precise nature of which is under further investigation.

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